

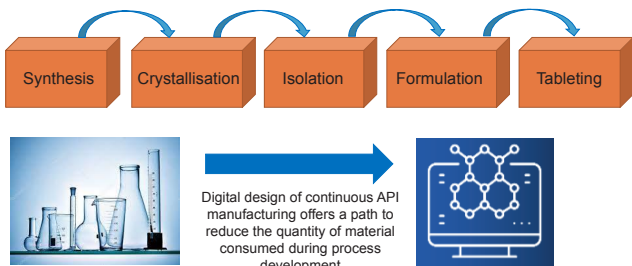


# Integrated Filtration and Washing modelling: Optimization of Impurity Rejection for the Filtration and Washing of Active Pharmaceutical Ingredients

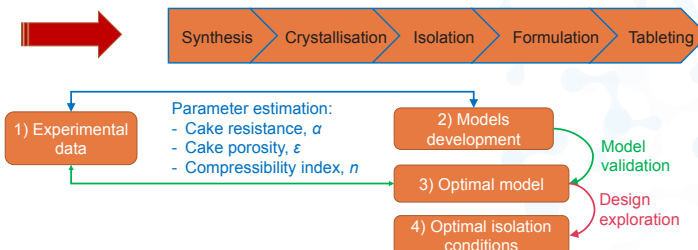
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**Context:** Transitioning pharmaceutical manufacturing from batch to continuous provides opportunity to improve sustainability.



**Aim:** Develop an integrated mechanistic filtration and washing model and accompanying optimization workflow to minimize impurities in the isolated filter cake.



## Materials and methods

Related impurities: 2-chlorobenzoic acid 2-3-dimethyl-N-phenylaniline Benzoic acid Copper (II) acetate  
 Filtration: 50mL dose stopped to dryland

Expt ID	PE or V	Crystallization Solvent	Wash Solvent	Isolation Pressure (mbar)	Wash Solvent cake volume	Number of Washes
1	PE	Ethyl Acetate	Cyclohexane	100	2	3
2	PE	Diglyme-Water	Heptane	600	2	3
3	PE	Ethyl Acetate	Heptane	600	2	2
4	PE	Ethyl Acetate	Heptane	100	4	2
5	V	Diglyme-Water	Cyclohexane	100	4	2
6*	PE	Diglyme-Water	Cyclohexane	350	3	3
7*	V	Diglyme-Water	Cyclohexane	350	3	3
8	V	Diglyme-Water	Heptane	100	4	3
9*	V	Diglyme-Water	Cyclohexane	350	3	3

Equipment/simulation tools	Purpose
Biotech VacMaster	Manual filtration and washing unit
Mastersizer 3000 laser diffraction	Particle size analyzer of the feed and isolated solid
HPLC	- Composition of filtrate removed during filtration and washing stages - Solid composition of the isolated cake
COSMOTerm	Solubility simulation
gPROMS FormulatedProducts	Filtration and washing process modelling

PE = experiments used for parameter estimation  
 V = experiments used for model validation  
 \* = experiment replicas

## Model development

- Filtration model (batch pressure filter)**
  - Carman – Kozeny equation<sup>(1)</sup> to simulate cake resistance
  - Darcy's law<sup>(2)</sup> filtration to simulate filtrate flow rate
  - Filtration stopped at dryland
- Integrated filtration and washing model (continuous pressure filter)**
  - Filtration simulated as in stage 1, integrated with washing
  - Washing modelled with displacement mechanism
  - No changes in solid phase are considered (no particle dissolution or growth)
- Washing model (MSMPR crystallizer)**
  - Filtration model (stage 1) outcome to feed washing model
  - Washing modelled with diffusion – dispersion mechanism<sup>(3)</sup>
  - No particle dissolution or growth

Factor	Definition	Unit	Factor	Definition	Unit
$\epsilon$	Cake porosity	-	$c$	Dry cake mass per unit volume of filtrate	kg m <sup>-3</sup>
$V_s$	Volume of solids	m <sup>3</sup>	$V$	Filtrate volume removed	m <sup>3</sup>
$V_{av}$	Total volume of cake	m <sup>3</sup>	$W$	Wash ratio	m <sup>3</sup> m <sup>-3</sup>
$\alpha_{av}$	Average specific cake resistance	m kg <sup>-1</sup>	$V_w$	Volume of wash	m <sup>3</sup>
$D_s$	Density of solids	kg m <sup>-3</sup>	$V_v$	Volume of voids	m <sup>3</sup>
$x_w$	Volume equivalent diameter	m	$u_w$	Superficial velocity of wash	m s <sup>-1</sup>
$z$	Time	s	$L$	Cake height	m
$A$	Filter area	m <sup>2</sup>	$c_{i,j}(t)$	Solute concentration initial, at the exit of the filter cake	-
$\Delta P$	Pressure drop along the filter axis	kg m <sup>-2</sup>	$c_{i,j}(w)$	Inlet wash solvent concentration	-
$\mu$	Filtrate viscosity	kg m <sup>-1</sup> s <sup>-1</sup>	$D_L$	Axial dispersion coefficient	m <sup>2</sup> s <sup>-1</sup>

## Results

### Parameter estimation

Crystallization Solvent	Wash Solvent	Cake Porosity (empirical value 0.44)	Simulated Medium Resistance (1/m)	Compressibility Index, n
Diglyme-water	Heptane	0.694	1.31x10 <sup>8</sup>	0.833
Diglyme-water	Cyclohexane	0.5258	1.31x10 <sup>7</sup>	0
Ethyl acetate	Heptane	0.4804	1.6x10 <sup>9</sup>	1.312
Ethyl acetate	Cyclohexane	0.476	1.46x10 <sup>9</sup>	0

- Pharmaceutical cakes are generally low-moderately compressible ( $n < 1$ ).<sup>(4)</sup>
- Cake and filtration estimated parameters show good fit with the porosity.
- Cyclohexane cases show  $n=0$  due to not enough data for PE.

### Model validation: washing models (Exp 3)

Solvent mass left filtration (kg)	Solvent mass left wash 1 (kg)	MFA concentration wash 1 (kg/kg)	CBA concentration wash 1 (kg/kg)
3.73E-03	9.56E-04	1.21E-03	2.48E-03
Experimental data			
2.06E-03	1.63E-03	1.47E-04	1.33E-04
Displacement model			
Diffusion dispersion model			
N/A	4.65E-04	1.11E-03	1.00E-03

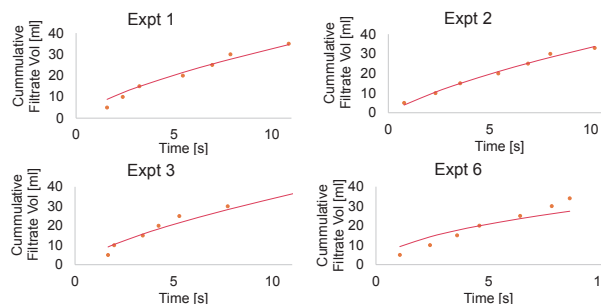
#### WASHING DISPLACEMENT MODEL:

- Accurate filtration end point simulated.
- Not able to predict the liquid composition due to lack of diffusion dilution mechanisms.

#### WASHING DIFFUSION DISPERSION MODEL:

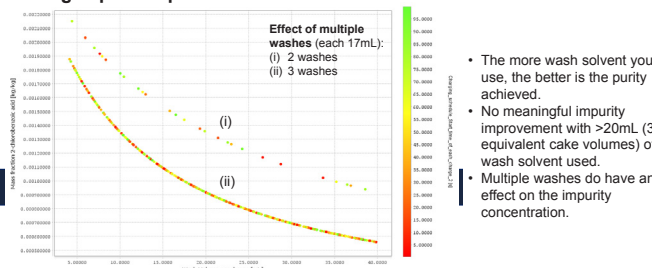
- Better accuracy in simulating the concentration of solute species removed during washing.
- Not able to predict residual solvent mass due to semi-batch operation and holdup specifications used.

### Model validation: filtration model



- Simulated Darcy's plots accurate in reproducing the experimental filtration flow rate evolution.
- Less accuracy for experiment 6: errors in manually collecting the experimental data.

### Design space exploration



- The more wash solvent you use, the better is the purity achieved.
- No meaningful impurity improvement with >20mL (3 equivalent cake volumes) of wash solvent used.
- Multiple washes do have an effect on the impurity concentration.

## Summary & Conclusions

- Developed a mechanistic model-based workflow for the optimization of an integrated filtration and washing model.
- Good match achieved between the estimated and experimental cake and filtration parameters.
- The pressure filter model is not able to predict the composition of filtrate removed well enough due to the displacement washing mechanism; better prediction achieved with diffusion-dispersion mechanisms, so diffusion-dispersion model used for design space exploration to identify which washing conditions reduce the final cake impurity concentration.
- Strong correlation between wash solvent volume used and number of washes and purity achieved.

## References

(1) Jin, Y., Zhu, Y. B., Li, X., Zheng, J., L., Dong, J., B., 2015. Trans. Porous Media, 433-453. (2) Beckmann W. Crystallization Basic Concepts and Industrial Applications, WILEY-VCH Verlag GmbH & Co. KGaA/2013 / Tien, C., 2002. Powder Technol., 127, 1-8. (3) Wakeman, R., J., Sabri, M., N., Tarterton, E., S., 1991. Powder Technol., 65, 283-292. (4) Stamatakis, K., Tien, C., 1991. Chem. Eng. Sci., 1917-1933. / Huhtanen, M., Salmiries, R., Kinnunen, T., Häkkinen, A., Elberg, B., Kallas, J., 2012. Sep. Sci. and Technol., 47, 1102-1112. (5) Tiller, F., M., Kwon, J., H., 1996. AIChE J., 2159.

